SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044. B.Sc. END SEMESTER EXAMINATIONS APRIL-2022 SEMESTER - I 20UCHCT1002 - General Chemistry II

Total Duration : 3 Hrs.

Total Marks : 60

Section A

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. Explain Bohr's model of an atom. Give its application and limitations.
- 2. Describe Baeyer's Strain theory.
- 3. Explain Schotty and Frenkel defects.
- 4. Distinguish between Protic and Aprotic solvents.
- 5. Write short note on a) Heisenberg's uncertainty principle. b) Aufbau's principle.
- 6. Describe Miller indices. How will you apply for BCC?
- 7. Explain the principles of qualitative analysis and illustrate with example.
- 8. What is ionisation potential? Give its factors and application.

Section B

Answer any **THREE** questions $(3 \times 10 = 30 \text{ Marks})$

- 9. a) Write Schrodinger's wave equation.
 - b) Describe and illustrate different quantum numbers.
- 10. a) Describe in detail about s, p, d and f block elements.
 - b) Explain Sanderson's electron density ratio.

11. Prepare the following reactions:

- a) Wurtz reaction. b) Kolbe's electrolytic method.
- c) Corey House method. d) Dickmann's ring closure
- e) Mechanism of halogenation.
- 12. a) What are the different types of unit cell which occur in crystals?
 - b) Write down the Born-Lande equation and define the various terms used.
 - c) Sketch and explain the structure of NaCl.
- 13. a) Describe Arrhenius theory.
 - b) Write note on HSAB principle.
 - c) What do you mean by common ion effect.
