22UPHAT1001

SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044.

B.Sc. END SEMESTER EXAMINATIONS NOVEMBER-2022

SEMESTER - I

22UPHAT1001 - Chemistry for Physics I

Total Duration : 2 Hrs 30 Mins.

Total Marks : 60

Section A

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. Explain briefly about the storage and handling of chemicals.
- 2. Decribe the principle of titrations.
- 3. Classify batteries and explain in detail.
- 4. Explain the composition and uses of the following fuels:
 - i) Bio-gas
 - ii) Water gas
 - iii) producer gas
- 5. Describe the structure of NaCl and NH_3 .
- 6. Classify various concentration terms and explain any two of it.
- 7. Sketch and explain about the principle and working of calomel electrode.
- 8. In what parameter the quality of fuels are examined? Justify- among solid, liquid and gaseous fuel, gaseous fuel is good? Why?

Section B

Answer any **THREE** questions $(3 \times 10 = 30 \text{ Marks})$

- 9. Explain the principle and applications of column chromatography. Describe how errors in chemical analysis can be eliminated and minimized?
- 10. Dicuss the applications of chromatographic technoies. Classify hydrogen bonding. Give its importance in biological systems.
- 11. Distinguish between primary and seconday standards. Outline acid-base titrations.
- 12. Define buffer. Obtain an expression to relate pH of an acidic buffer to the concentration of salt and acid. How does Li-ion batteries organized? Explain its electrode reactions.
- 13. Differentiate between nuclear fission and fusion reactions. Appraise applications of radio isotopes in medicine and agriculture fields.
