22PCHCT1003

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M.Sc.(Chemistry) - END SEMESTER EXAMINATIONS APRIL - 2023 SEMESTER - I

22PCHCT1003 - Chemical Kinetics and Thermodynamics

Total Duration: 2 Hrs. 30 Mins. Total Marks: 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. Describe collision theory of rate of reaction.
- 2. Show the variation of chemical potential with respect to temperature and pressure.
- 3. Derive the relation between partition function and thermo dynamic function.
- 4. Describe about the relations for rotational partition function.
- 5. Give a brief note on the rate of an enzyme catalyzed reaction.
- 6. Derive Gibbs-Duhem equations.
- 7. Derive and discuss the Debye theory of heat capacities of solids.
- 8. Explain the Application of Bose-Einstein statistics.

Section C

I - Answer any **TWO** questions $(2 \times 10 = 20 \text{ Marks})$

- 9. Describe the flash photolysis methods for studying kinetics of fast reactions.
- 10. Discuss the Langmuir Hinshelwood mechanism for biomolecular surface reactions.
- 11. a) What is Onsager reciprocal relation?
 - b) Prove the Onsager reciprocal relation by the principle of microscopic reversibility.
- 12. Discuss on partial molar volume and partial molar heat content. Explain its significance.

II - Compulsory question $(1 \times 10 = 10 \text{ Marks})$

13. Derive the following:

a) Sackur-Tetrode equation (5)

b) Molecular translational partition function (5)
