#### 22PCHCT2005

# SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS)

(Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044.

M.Sc.(Chemistry) - END SEMESTER EXAMINATIONS APRIL - 2023 SEMESTER - II

22PCHCT2005 - Coordination Chemistry: Bonding, Reaction and Spectra

Total Duration: 2 Hrs. 30 Mins. Total Marks: 60

### Section B

Answer any **SIX** questions  $(6 \times 5 = 30 \text{ Marks})$ 

- 1. Illustrate the shapes and magnetic property of four coordinated complexes of  $Ni^{2+}$  with reference to VBT.
- 2. Apply pi-bonding theory and explain the trans effect.
- 3. Deduce the possible term symbols for  ${\sf Ti}^{2+}$  and mention the microstates in this system.
- 4. Discuss the vibrational spectra of metal carbonyls.
- 5. Apply SN<sup>1</sup> mechanism and show that octahedral complexes undergo dissociation via five coordinated intermediate.
- 6. Assess the effect of  $\pi$  bonding on the magnitude of  $\Delta$ .
- 7. Compare Orgel and Sugano- Tanabe diagram.
- 8. Analyse the effect of Jahn Tellar distortion and spin orbit coupling on electronic spectra of complexes.

## Section C

I - Answer any **TWO** questions  $(2 \times 10 = 20 \text{ Marks})$ 

- 9. Explain  $\pi$  bond formation of a metal ion with PR<sub>3</sub> ligand on the basis of MO theory.
- 10. Assess the mechanism of acid hydrolysis reactions of octahedral complexes with suitable examples.
- 11. How would you use the group vibrations in the structural elucidation of metal complexes of cyanide, sulphate, & thiocyanates and the effect of isotopes on the vibrational spectra of molecule.
- 12. Explain the electronic transitions of Cr(III) complexes based on Sugano- Tanabe diagram.
  - II Compulsory question  $(1 \times 10 = 10 \text{ Marks})$
- 13. Analyze the outer sphere mechanism of octahedral complexes with two examples.

\*\*\*\*\*

#### 22PCHCT2005

# SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS)

(Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044.

M.Sc.(Chemistry) - END SEMESTER EXAMINATIONS APRIL - 2023 SEMESTER - II

22PCHCT2005 - Coordination Chemistry: Bonding, Reaction and Spectra

Total Duration: 2 Hrs. 30 Mins. Total Marks: 60

## Section B

Answer any **SIX** questions  $(6 \times 5 = 30 \text{ Marks})$ 

- 1. Illustrate the shapes and magnetic property of four coordinated complexes of  $Ni^{2+}$  with reference to VBT.
- 2. Apply pi-bonding theory and explain the trans effect.
- 3. Deduce the possible term symbols for  ${\sf Ti}^{2+}$  and mention the microstates in this system.
- 4. Discuss the vibrational spectra of metal carbonyls.
- 5. Apply SN<sup>1</sup> mechanism and show that octahedral complexes undergo dissociation via five coordinated intermediate.
- 6. Assess the effect of  $\pi$  bonding on the magnitude of  $\Delta$ .
- 7. Compare Orgel and Sugano- Tanabe diagram.
- 8. Analyse the effect of Jahn Tellar distortion and spin orbit coupling on electronic spectra of complexes.

## Section C

I - Answer any **TWO** questions  $(2 \times 10 = 20 \text{ Marks})$ 

- 9. Explain  $\pi$  bond formation of a metal ion with PR<sub>3</sub> ligand on the basis of MO theory.
- 10. Assess the mechanism of acid hydrolysis reactions of octahedral complexes with suitable examples.
- 11. How would you use the group vibrations in the structural elucidation of metal complexes of cyanide, sulphate, & thiocyanates and the effect of isotopes on the vibrational spectra of molecule.
- 12. Explain the electronic transitions of Cr(III) complexes based on Sugano- Tanabe diagram.
  - II Compulsory question  $(1 \times 10 = 10 \text{ Marks})$
- 13. Analyze the outer sphere mechanism of octahedral complexes with two examples.

\*\*\*\*\*