22UCHCT1001

SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS)

(Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044.

B.Sc.(Chemistry) END SEMESTER EXAMINATIONS NOVEMBER -2023

SEMESTER - I

22UCHCT1001 - Basic Concepts in Inorganic Chemistry

Total Duration : 2 Hrs 30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. State Paulis exclusion principle. Indicate the usefulness of the principle using an example.
- 2. What are the limitations of Bohr's theory?
- 3. Lanthanides and Actinides are placed in separate rows at the bottom of the periodic table. Explain the reason for this arrangement.
- 4. What is the pauling and Mulliken scale of electronegativity?
- 5. Explain the coordinate bond formation between BF_3 and NH_3 .
- 6. Calculate the lattice energy of NaCl using Born-Haber cycle.
- 7. Discuss in brief on packing of ions in crystals.
- 8. Write a note on
 - i) Common ion effect and ii) Solubility Product

Section C

Answer any **THREE** questions $(3 \times 10 = 30 \text{ Marks})$

- 9. Explain various quantum numbers which completely specify the electron of an atom.
- 10. Explain the following terms.
 - a) Ionization potential b) Electron affinity c) Electro negativity d) Inert pair effect e) Covalent radii
- 11. Draw the Crystal Structure of zinc blende and cesium chloride and explain.
- 12. Write short notes on
 - a) Schottky defect b) Frankel defect.
- 13. Discuss the following with suitable examples.
 - a) Lewie theory of acids and bases
 - b) HSAB principle
 - c) Usanovich theory of acids and bases.

(4+2+4)
