SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044. B.Sc.(Chemistry) END SEMESTER EXAMINATIONS NOVEMBER -2023 SEMESTER - I 22UCHCT1002- Analytical Chemistry - I

Total Duration : 2 Hrs 30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. How is the handling and storage of flammable and corrosive Chemicals done at laboratory?
- 2. List out different methods for the minimization of systematic errors and explain.
- 3. What is Complexometric titration? How to determine the hardness of water by using EDTA.
- 4. What is masking agent? Explain with suitable examples and give its uses.
- 5. What are the factors affecting degree of ionization.
- 6. What are interfering anions? How are they elimiated?
- 7. Explain the characteristics of a precipitate and precipitating agent.
- 8. a) Discuss about common ion effect.
 - b) Define Buffer solutions.

Section C

Answer any **THREE** questions $(3 \times 10 = 30 \text{ Marks})$

- 9. a) Discuss the general precautions for avoiding accidents.
 - b) Explain the disposed of toxic waste and poisons.
- 10. a) "Precision always accompanies accuracy but high degree of precision does not mean accuracy" Justify the statement.
 - b) The normality of a solution is determined by four separate titrations and the results are 0.2041, 0.2049, 0.2039 and 0.2043. Calculate the Mean, Median, Range, Standard deviation and coefficient of variation.
- 11. a) What is a primary and secondary standard? What requirements should be satisfied as a primary standard?
 - b) Discuss any one theory of indicators.

(5+5)

- 12. a) What is meant by Post precipitation? How is it differ from co-precipitation.
 - b) Discuss with suitable example for solubility and ionic product.
- 13. Write any one test for the following anions and cations.
 - a) Phosphate b) borate c) Copper d) Calcium

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