SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044. B.Sc.(Physics) END SEMESTER EXAMINATIONS NOVEMBER -2023 SEMESTER - I **22UPHAT1001 - Chemistry for Physics I**

Total Duration : 2 Hrs 30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. List out the general rules to be followed in storage and handling of chemicals.
- 2. Define the following (i) Molarity (ii) Molality (iii) ppm.
- 3. Define buffer solutions? Derive Henderson-Hasselbalche equation and write its significance.
- 4. Describe the requisites of a good fuel? Write down the different types of fuels.
- 5. Explain the structure of NaCl. Write the characteristics of ionic compounds.
- 6. Highlight the importance and preparation of primary and secondary standards.
- 7. Explain the following:(i) EMF (ii) Standard electrode potential (iii) Reference electrode.
- 8. Write about (i) Octane number (ii) TEL (iii) CNG

Section C

Answer any **THREE** questions $(3 \times 10 = 30 \text{ Marks})$

- 9. (a) Explain the principle of thin layer chromatography. How is it performed?
 - (b) Describe with a diagram, how a substance can be purified by steam distillation.
- 10. (a) Predict the hybridization of (i) CH_4 (ii) C_2H_4
 - (b) Discuss about the types and importance of hydrogen bonding.
- 11. Explain the classification of different types of acid-base titrations and their principle.
- 12. What are primary and secondary batteries? Explain in detail about any two types of batteries.
- 13. (a) Differentiate between nuclear fission and fusion reactions. (5+5)(b) Highlight the uses of radioisotopes in our life.
