22PPHET3003

SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044. M.Sc.(Physics) END SEMESTER EXAMINATIONS NOVEMBER - 2023 SEMESTER - III 22PPHET3003 - Crystal Physics

Total Duration : 2 Hrs. 30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

1. What is heterogeneous nucleation? Explain.

2. Explain the slow cooling method of crystal growth technique.

- 3. Describe the salient features of covalent and metallic bondings. Give examples.
- 4. Explain the concept of reciprocal lattice.
- 5. Explain Vickers hardness test to determine the hardness number of a crystal.
- 6. Briefly explain about disc and cap shaped nucleus.
- 7. Obtain an expression for solubility of super saturation solution.
- Determine the glancing angle on the cube (1 1 0) of a rock salt crystal (a=0.2814 nm) corresponding to second order diffraction maximum for the X-ray of wavelength 0.071 nm.

Section C

I - Answer any **TWO** questions $(2 \times 10 = 20 \text{ Marks})$

- 9. Derive Gibbs Thompson equation for vapour.
- 10. Describe the principle and structure of gel growth technique for crystallization.
- 11. What are ionic crystals? Explain the formation of an ionic crystal and obtain an expression for its cohesive energy.
- 12. Draw the block diagram of a FTIR spectrometer and explain its instrumentation and working.

II - Compulsory question $(1 \times 10 = 10 \text{ Marks})$

13. Discuss the different steps in the analysis of crystal structure and explain how it helps in structural determination.
