22PCHCT1003

SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS)

(Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai — 600 044.

M.Sc.(Chemistry) END SEMESTER EXAMINATIONS NOVEMBER - 2023 SEMESTER - I

22PCHCT1003 - Chemical Kinetics and Thermodynamics

Total Duration : 2 Hrs. 30 Mins.

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

Section B

- 1. Brief the role of continuous flow and stopped flow techniques with neat block diagrams in studying fast reactions.
- 2. Relate the reversible, consecutive and parallel reactions with relevant examples.
- 3. Predict the types of mechanisms involved in the inhibition of enzyme catalyzed reactions.
- 4. Explain the variation of chemical potential with temperature and pressure.
- 5. Describe the determination of fugacity by graphical method with a neat sketch.
- 6. Write a comparative account on the formation of HCI & HBr. Mention the explosion limit of reactions.
- 7. Show the study of phenomenological approach in Onsager equation.
- 8. Recommend an expression for entropy of ideal gases as proposed by Sakur Tetrode.

Section C

I - Answer any **TWO** questions $(2 \times 10 = 20 \text{ Marks})$

- 9. (i) Describe in detail the estimation of free energy, entropy and enthalpy of activation and highlight their significance (6 marks)
 - (ii) Interpret the polar and steric sensitivity factors of Taft equation. (4 marks)
- 10. Apply Hinshelhood and Lindemann theories in predicting the unimolecular reactions
- 11. Examine the following in detail
 - (i) Debye theory (5 marks)
 - (ii) Dulong and Petit's law
- 12. Compare Maxwell Boltzmann, Fermi Dirac and Bose Einstein Statistics elaborately.

II - Compulsory question $(1 \times 10 = 10 \text{ Marks})$

13. Examine about concepts of activity and activity coefficients in a vivid manner.

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Total Marks : 60

(5 marks) (5 marks)