22UCHCT2004

SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai - 600 044. B.Sc.Chemistry - END SEMESTER EXAMINATIONS - NOV'2024 SEMESTER -II 22UCHCT2004-Analytical Chemistry -II

Total Duration : 2 Hrs.30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. What is the significance of Faraday's laws of electrolysis in electrodeposition.
- 2. Illustrate how ilkovic equation governs polarographic measurement and recognize the terms involved.
- 3. Discuss the physical/chemical phenomena causing changes in temperature in TGA.
- 4. Distinguish between static and dynamic TGA.
- 5. How is fractional crystallization used to separate a mixture of solid compounds.
- 6. List the factors influencing Rf value in thin layer chromatography. What are the advantages of TLC over the other chromatographic techniques.
- 7. Explain ascending and descending development methods used for the separation of component in mixture by paper chromatography with neat diagram.
- 8. Explain about the detectors used to measure the physical property of the solute in HPLC.

Section C

Answer any **THREE** questions $(3 \times 10 = 30 \text{ Marks})$

- 9. How is polarography used to determine trace metals in solutions.
- What is a Soxhlet extractor, and how does it differ from simple solvent extraction? Explain with illustrative diagram.
- 11. Infer and illustrate the various steps involved in a thermogravimetric graph.
- 12. Discuss the different types of paper chromatography and compare their merits and demerits.
- 13. Examine the process involved in the plate theory of gas chromatography.
