22UCHCT3006

SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai - 600 044. B.Sc.Chemistry - END SEMESTER EXAMINATIONS - NOV'2024 SEMESTER - III 22UCHCT3006 - Chemistry of s and p Block Elements

Total Duration : 2 Hrs.30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. What is meant by diagonal relationship? Analyse in what aspects does lithium resemble magnesium?
- 2. How dinitrogen tetroxide is prepared? Describe its nature.
- 3. Discuss the bonding and structure of SO_3 .
- 4. List few industrial applications of noble gases.
- 5. Explain the electron deficiency nature of boron compounds.
- 6. Describe the evidence for the statement -metallic character increases on going down the group 15.
- 7. The chemistry of oxygen is unique compared to those of the other members in its group. Validate this fact.
- 8. How does water react with XeF_2 , XeF_4 , and XeF_6 ?

Section C

Answer any **THREE** questions $(3 \times 10 = 30 \text{ Marks})$

- 9. Discuss the complexation tendencies of s block elements.
- 10. In detail explain the preparation and structural characteristics of silicones.
- 11. Describe the chemistry of the oxides of nitrogen.
- 12. Outline the principle of the contact process for the preparation of sulphuric acid.
- 13. What are Pseudo halogens? Analyze and explain the similarities with halogens and pseudohalogens.
