22UCHCT4007

SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai - 600 044. B.Sc.Chemistry - END SEMESTER EXAMINATIONS - NOV'2024 SEMESTER - IV 22UCHCT4007 - Thermodynamics

Total Duration : 2 Hrs.30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. Discuss the postulates of kinetic theory of gas.
- 2. Derive an expression which relates C_p and C_v .
- 3. Explain exothermic and endothermic reactions with examples.
- 4. Explain the heat of formation and heat of combustion reactions.
- 5. Define Fugacity. How will you determine fugacity by graphical method?
- 6. Discuss about the activity and activity co-efficient.
- 7. State the law of mass action and derive an expression for equilibrium constant.
- 8. Explain the Nernst heat Theorem.

Section C

Answer any **THREE** questions $(3 \times 10 = 30 \text{ Marks})$

- 9. Derive Vanderwaal's equation of state. (10)
 10. Describe the process of water calorimeter and bomb calorimeter. (5 + 5)
 11. a. Derive Gibb's-Helmholtz equation (5)
- b Derive Gibb's-Duhem equation (5)
- 12. State Le-Chatlier's principle. How does the formation of ammonia take place with the effect of change in temperature and pressure? (10)
- Determine the absolute entropy of a substance using third law of thermodynamics. (10)
