SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN (AUTONOMOUS) (Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC) Chromepet, Chennai - 600 044. M.Sc.Chemistry - END SEMESTER EXAMINATIONS - NOV' 2024 SEMESTER - I 22PCHCT1003 - Chemical Kinetics and Thermodynamics

Total Duration : 2 Hrs. 30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions $(6 \times 5 = 30 \text{ Marks})$

- 1. What is the difference between high and low activation energies and how this affects the reaction?
- 2. Explain any one method of fast reaction.
- 3. Derive the relationship between substrate and enzyme concentrations and the rates of enzyme-catalyzed reactions.
- 4. Explain the concept of Lindemann and Hinshelwood theory.
- 5. Explain Gibbs Duhem equation.
- 6. Discuss the terms Concepts of activity and activity coefficient.
- 7. Describe the transformation properties of fluxes and forces in a chemical reaction.
- 8. Derive the expression for Maxwell Boltzmann statistics.

Section C

I - Answer any **TWO** questions $(2 \times 10 = 20 \text{ Marks})$

- 9. Predict the kinetics of complex for the parallel reaction
- 10. Apply Nernst distribution law, under what condition is it valid? How is the law derived from thermodynamic considerations?
- 11. Justify an expression for the average energy of one of material oscillator according to Einstein model and obtain heat capacity at constant volume.
- 12. Determine an expression for the molecular vibrational partition function of an ideal diatomic gas.

II - Compulsory question $(1 \times 10 = 10 \text{ Marks})$

13. Justify and explain Rice–Herzfeld mechanism for the decomposition of acetaldehyde.
